A One-step Synthesis of 1,6-Naphthyridine

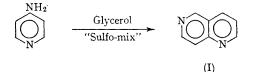
By THOMAS J. KRESS and WILLIAM W. PAUDLER (Department of Chemistry, Ohio University, Athens, Ohio, U.S.A.)

DURING the past ten years, several syntheses¹⁻³ of 1,6-naphthyridine (I) have been described. All of these preparations involve at least seven-step sequences yielding the 1,6-naphthyridine in overall yields of two per cent or less.

Our interest in naphthyridine chemistry⁴⁻⁶ prompted us to investigate methods of syntheses which would afford the naphthyridines in higher yields and in fewer steps. The most reasonable approach would be the synthesis *via* a Skraup reaction.

All literature describing the syntheses of 1,6naphthyridines (see ref. 7 and references therein) states that the Skraup synthesis is not applicable to 4-aminopyridine. The rationale for this assumption centres on the "low" basicity $(pK_a = 9\cdot 2)$ of the 4-aminopyridine (this statement was made prior to the availability of the pK_a data, which, of course, reveals 4-aminopyridine to be a fairly strong base). This conclusion is, however, not valid in view of the fact that 4-aminoquinoline $(pK_a = 9\cdot 2)$ undergoes reactions involving intermediates similar to those expected in the Skraup synthesis.^{7a-c}.

It became consequently of considerable interest to apply the Skraup reaction to 4-aminopyridine, utilizing the "sulfo-mix" described by Utermohlen.⁸ When this reaction was attempted, the parent 1,6naphthyridine (I) was obtained in forty per cent yield.



The reaction conditions employed in this preparation are the same as those described for the preparation of 1,5-naphthyridine by the Skraup method,² except for the use of "sulfo-mix" in place of the traditional oxidizing mixture. Anhydrous glycerol (25 g.) is added to 117 g. of cold $(0-5^{\circ})$ "sulfo-mix" (a mixture of nitrobenzenesulphonic acids in sulphuric acid) and to this mixture is added 7.5 g. (0.08 mole) of 4-aminopyridine followed immediately by the addition of 40 ml. of water. This mixture is stirred until homogeneous (ca. 10 min.) and is finally heated in an oil bath with vigorous stirring at 130° for 5 hr. The resulting mixture is made basic (with cooling, ice-salt) to pH 10 with concentrated (50%) aqueous sodium hydroxide. The resulting mixture is subjected to stream distillation until the distillate no longer forms a precipitate with picric acid (approximately 3 l. of distillate is obtained). Extraction of the distillate with a total of 2 l. of chloroform, and drying of the chloroform extract with anhydrous MgSO4, afforded, after removal of the solvent in vacuo, 4 g. of CHEMICAL COMMUNICATIONS

pure 1,6-naphthyridine. The properties of the 1,6naphthyridine thus obtained (m.p., n.m.r. spectrum, m.p. of the picrate) are identical to those reported¹⁻⁴ for 1,6-naphthyridine prepared by the classical multistep sequences.

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- ¹ N. Ikekawa, Chem. Pharm. Bull. (Tokyo), 1959, 6, 263.
- ² A. Albert, J. Chem. Soc., 1960, 1790.
 ³ B. Ferrier and N. Campbell, Proc. Roy. Soc. Edinburgh, Sect. A, 1959–1960, 65, 23.

- ⁶ B. Ferrier and N. Campbell, Proc. Roy. Soc. Earnburgh, Sect. A, 1959-1960, 65, 23.
 ⁴ W. W. Paudler and T. J. Kress, J. Heterocyclic Chem., 1966, 2, 393.
 ⁵ W. W. Paudler and T. J. Kress, Chem. and Ind., 1966, 13, 3055.
 ⁶ W. W. Paudler and T. J. Kress, Chem. and Ind., 1966, 1557.
 ⁷ (a) R. C. Elderfield, "Heterocyclic Compounds," John Wiley and Sons, Inc., New York, 1961, Vol. 7, pp. 198-236;
 (b) B. Bobranski and E. Sucharda, Ber., 1926, 60, 1081; (c) B. Brobanski and E. Sucharda, Roczniki Chem., 1927, 7, 192.
 ⁸ W. P. Utermohlen, Jr., J. Org. Chem., 1943, 8, 544.